

Supporting Information

Facile Resolution of Racemic Terbutaline and A Study of Molecular Recognition through Chiral Supramolecules Based on Enantiodifferentiating Self-Assembly

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1) ^1H and ^{13}C NMR analyses of (*R*)-1.4 and (*S*)-1.4

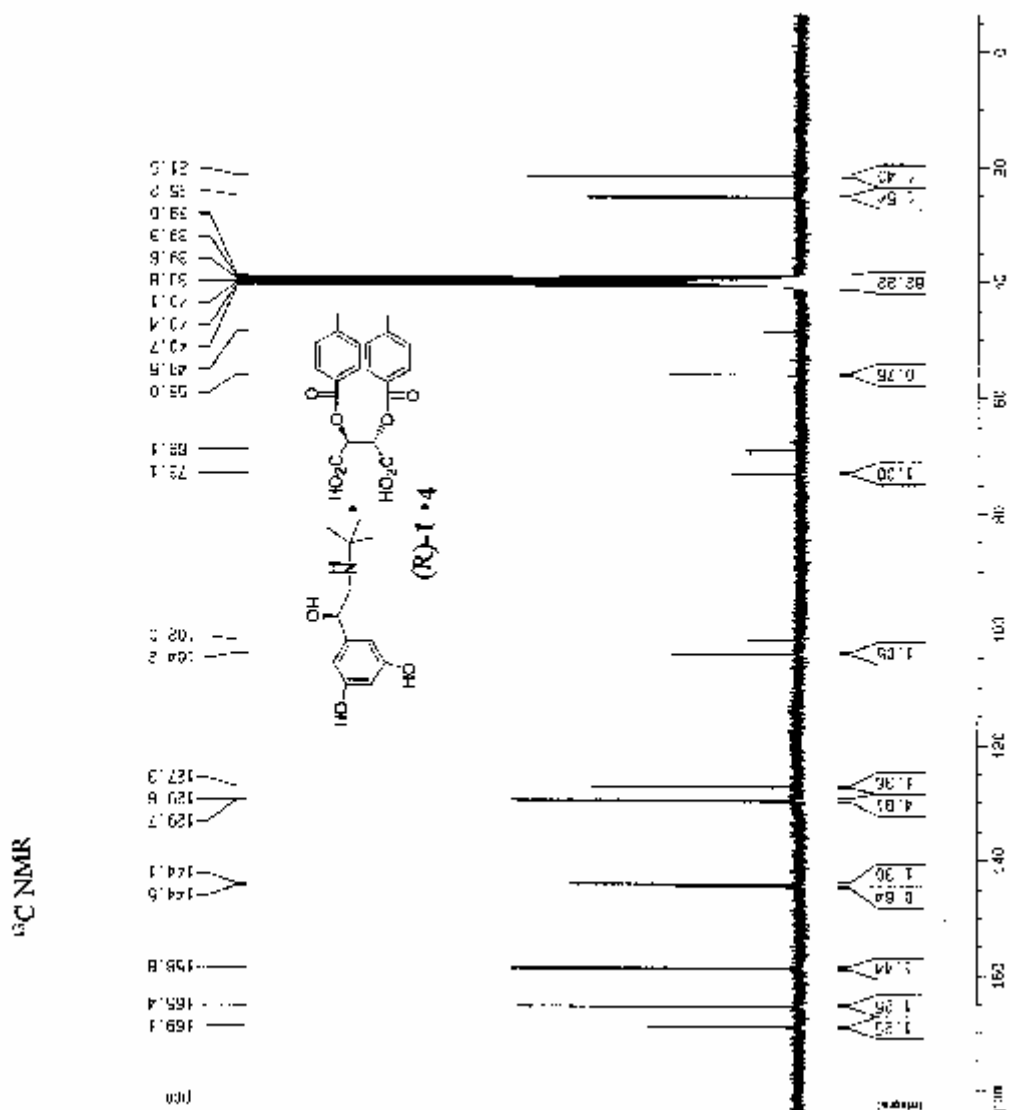


Figure S1. ^{13}C -NMR spectrum of (*R*)-terbutaline (1) and *D*-DTTA (4)

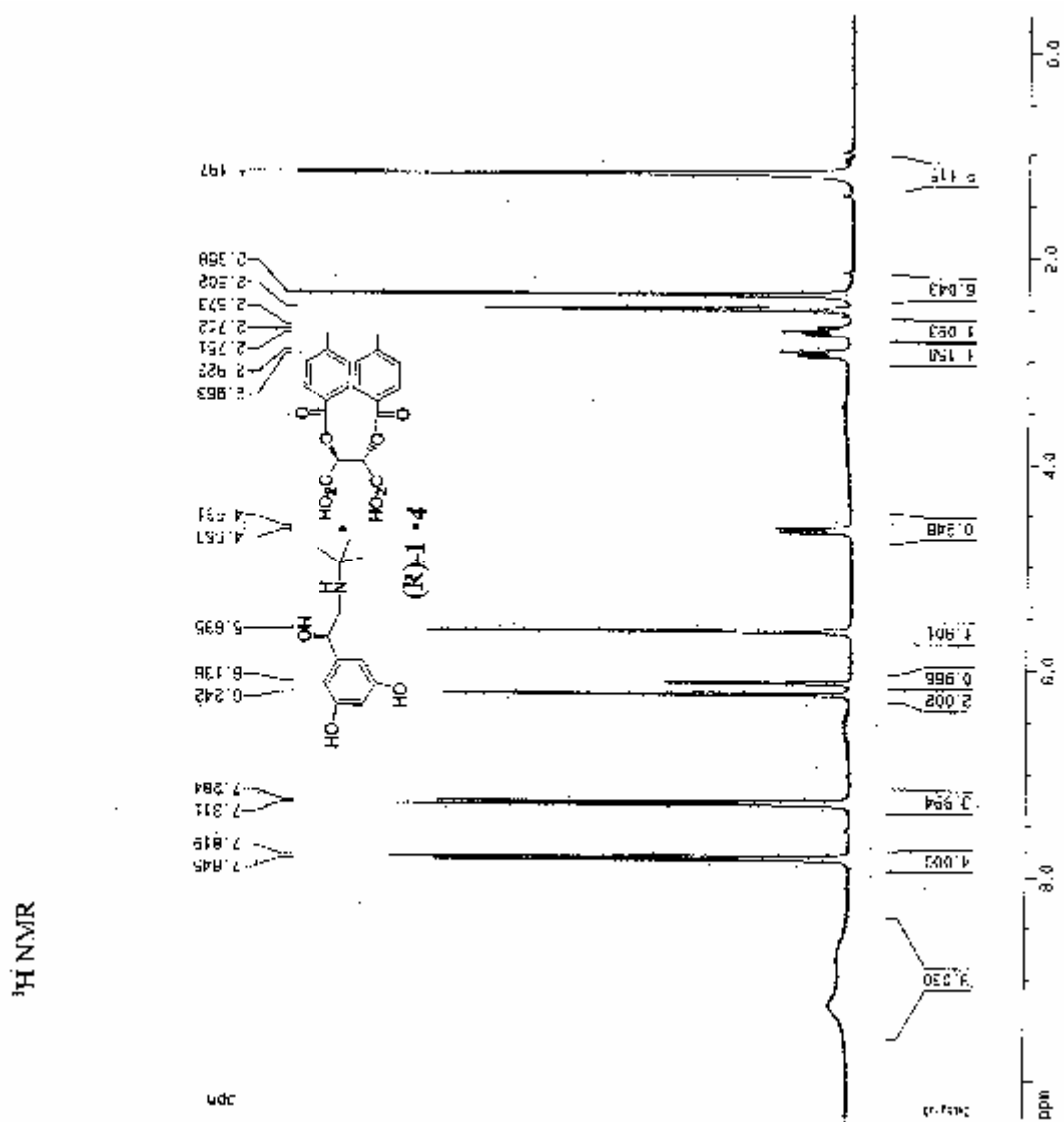


Figure S2. ¹H-NMR spectrum of (*R*)-terbutaline (1) and *D*-DTTA (4)

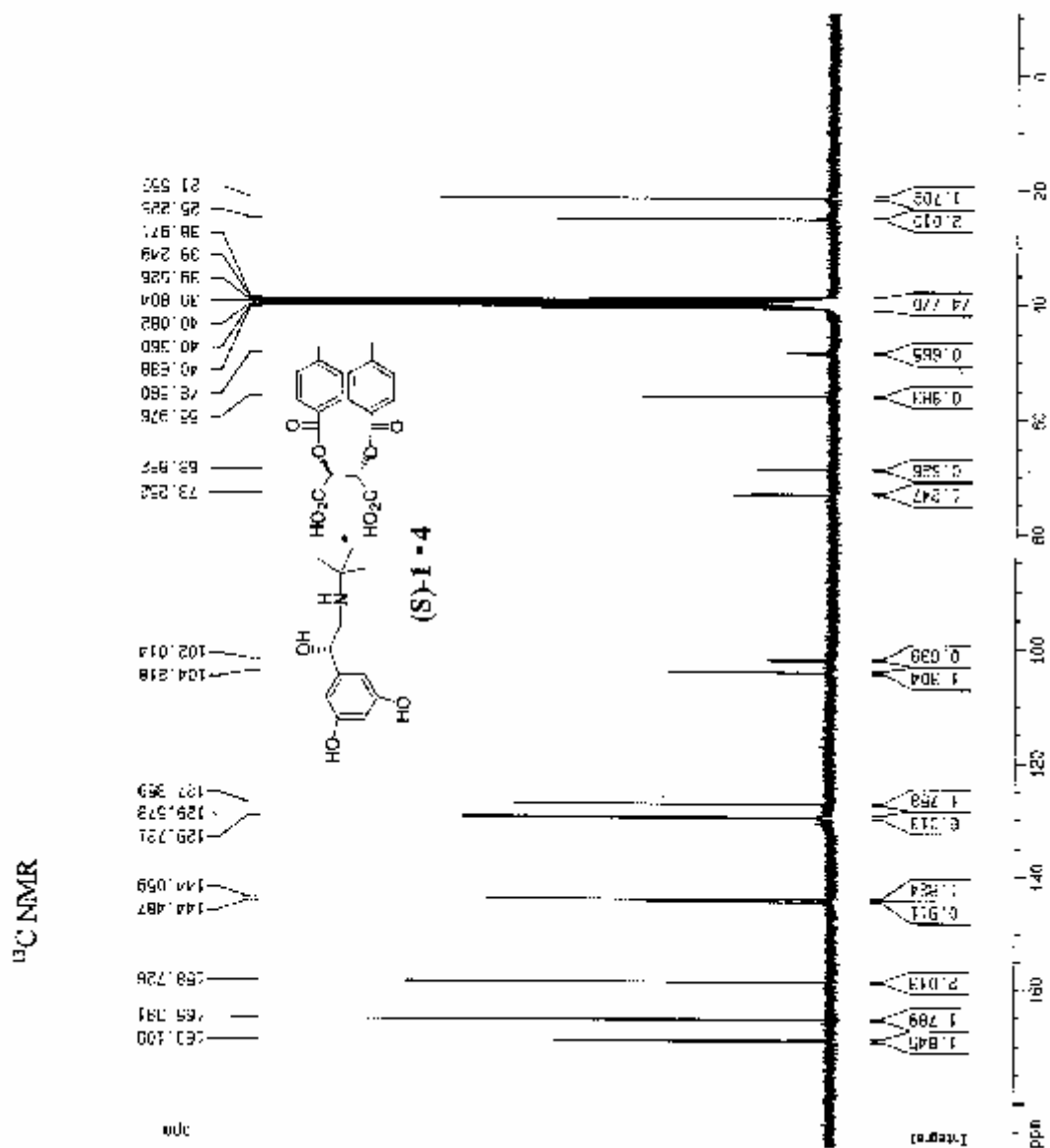


Figure S3. ¹³C-NMR spectrum of (*S*)-terbutaline (1) and *D*-DTTA (4)

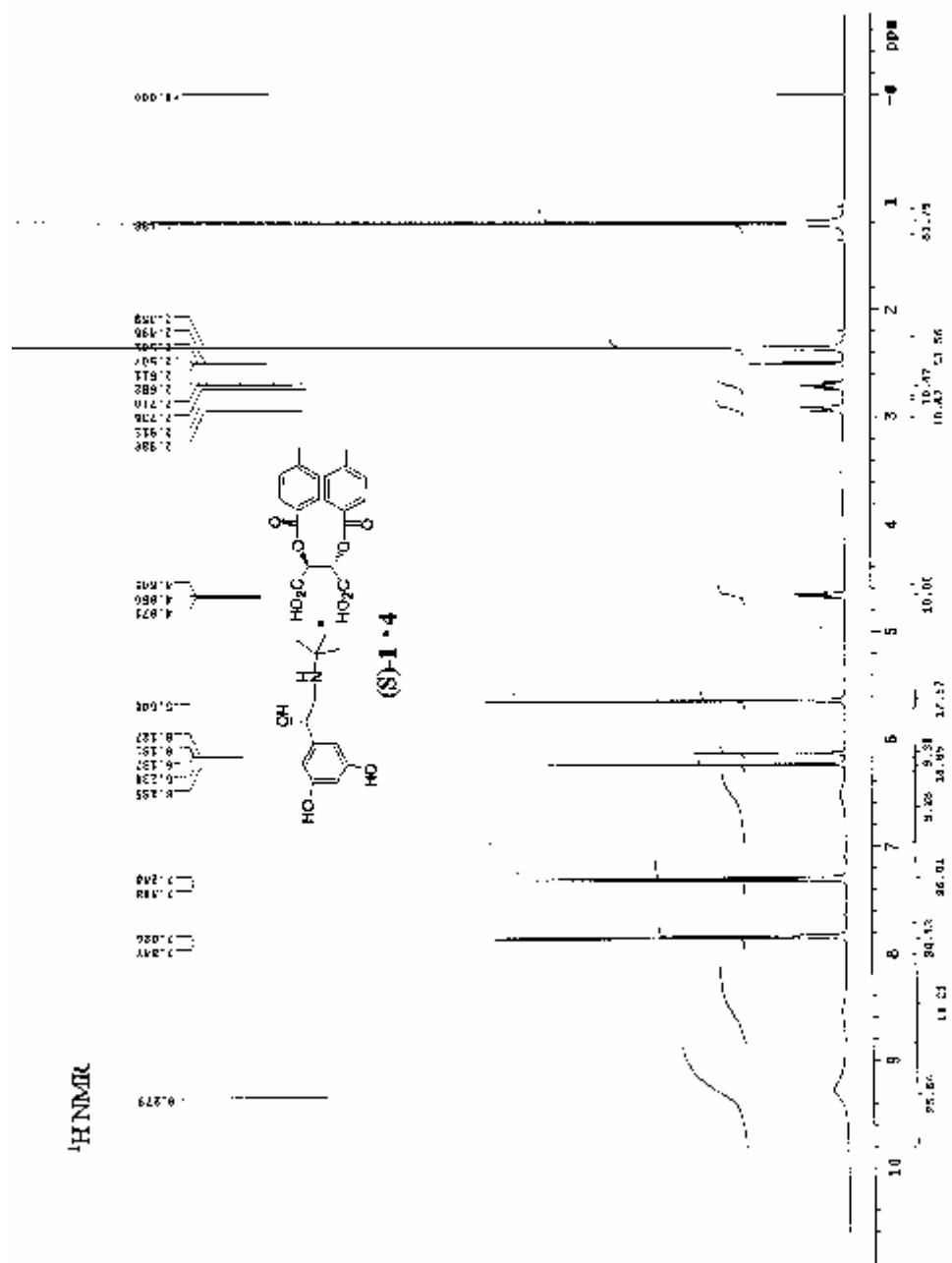


Figure S4. ¹H-NMR spectrum of (*S*)-terbutaline (1) and *D*-DTTA (4)

2) ORTEP views and stacking structures of crystals **5**, **7** and **8**

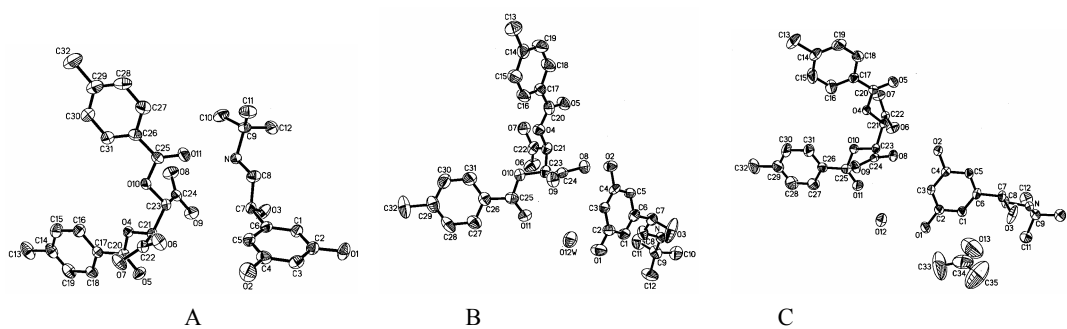


Figure S5. ORTEP views of the molecular structure of **5** (A), **7** (B) and **8** (C).

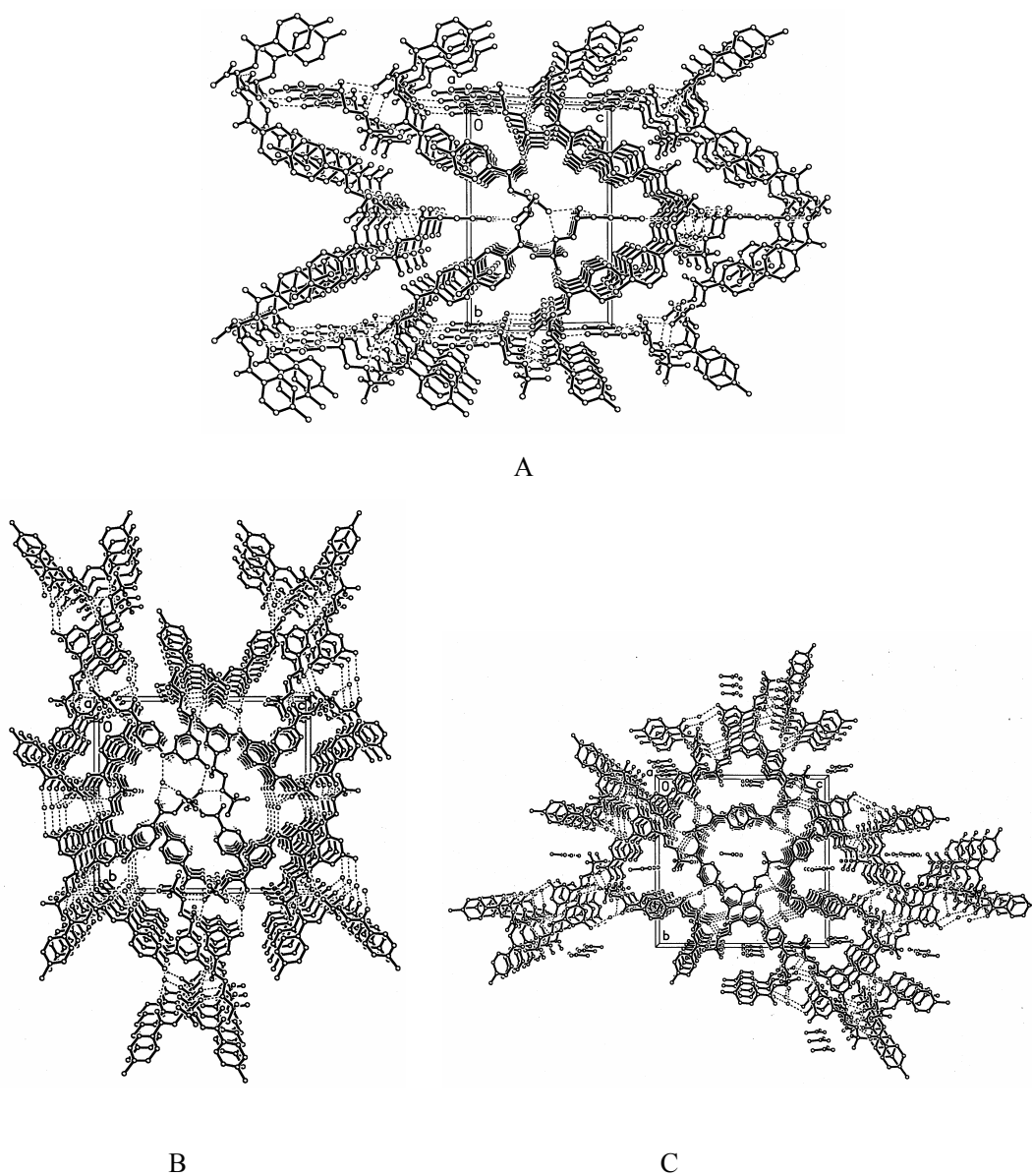


Figure S6. Stacking structures of crystals **5** (A), **7** (B) and **8** (C) (top view down the *a* axis). All H-bonds are represented by dotted lined.

3) DSC and TG analyses of the more- and less-soluble salts

A) For more-soluble salt

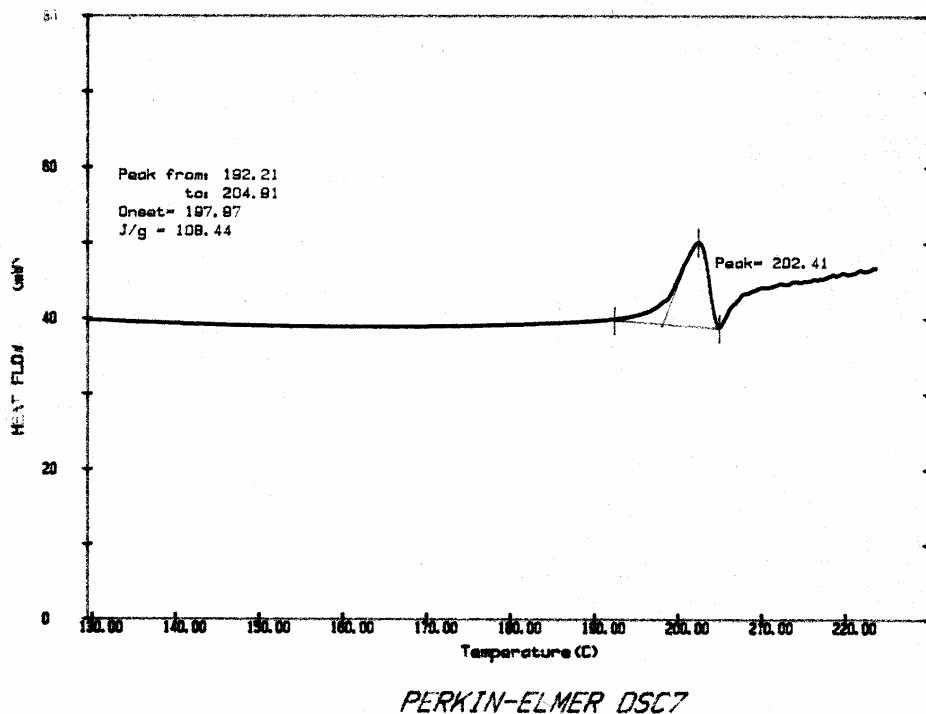


Figure S7. DSC analysis of the more-soluble salt. $\Delta H_{\text{fus}} = 108.44 \times 611/1000 = 66.2$ KJ/mol

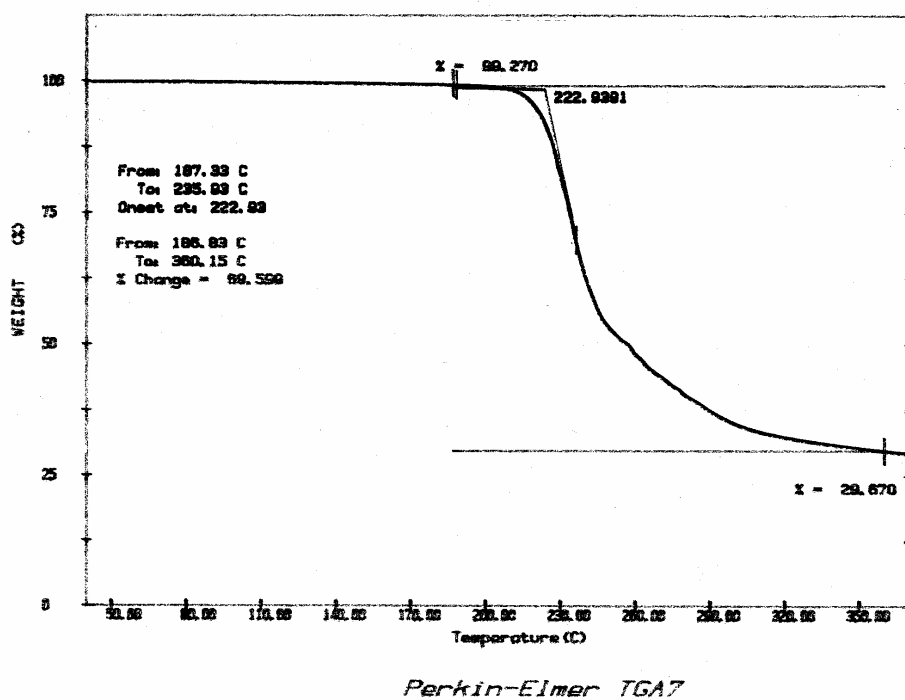


Figure S8. TG analysis of the more-soluble salt

B) For less-soluble salt

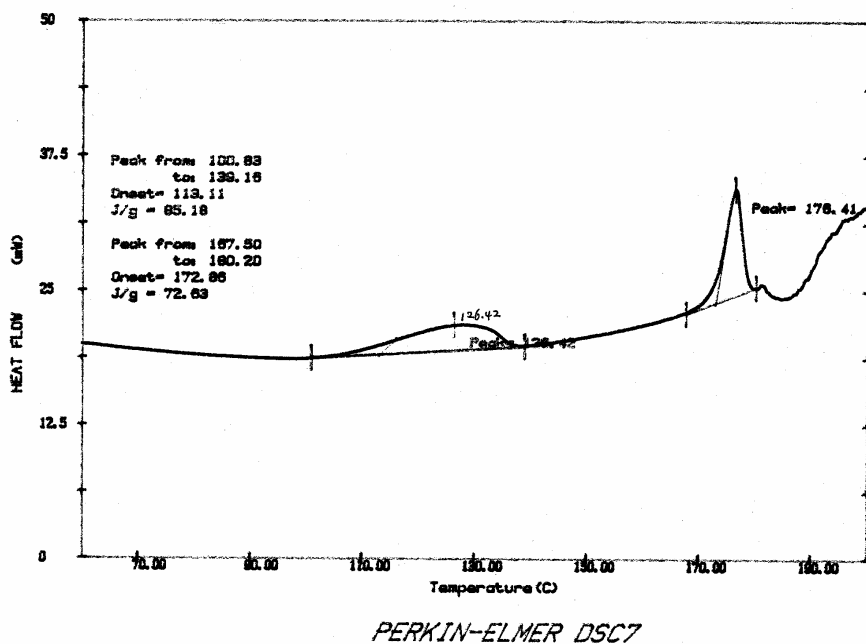


Figure S9. DSC analysis of the less-soluble salt. $\Delta H_{\text{fus}} = (85.18 \times 629/1000) + (72.63 \times 611/1000) = 53.6 + 44.4 = 98.0$ KJ/mol, the first curve is relative to water loss ($\Delta H = 53.6$ KJ/mol)

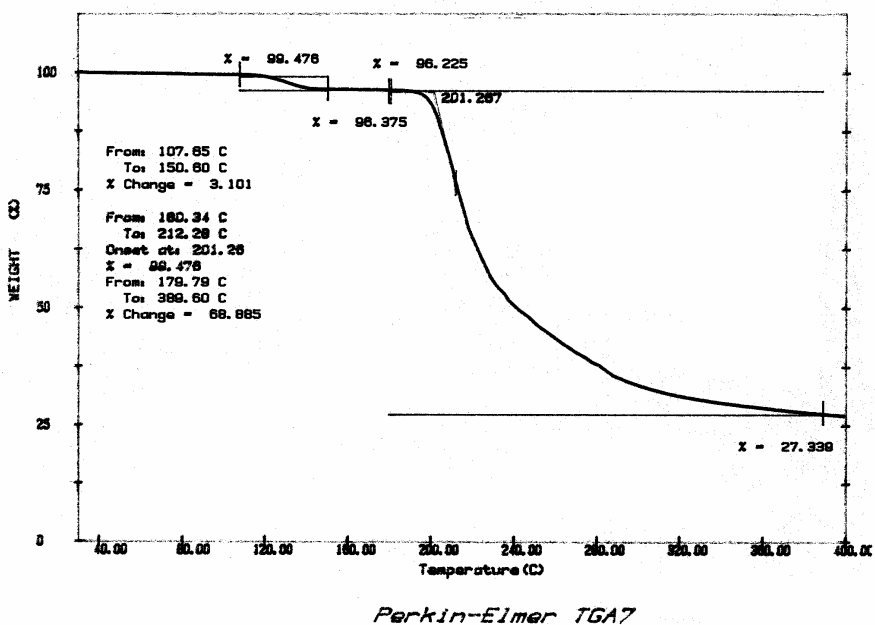


Figure S10. TG analysis of the less-soluble salt